

Radioactivity: beta negative particle



Ask Mish



A beta particle is identical to an electron (charge = -1).

Beta decay occurs in atoms with too many neutrons present in the nucleus.

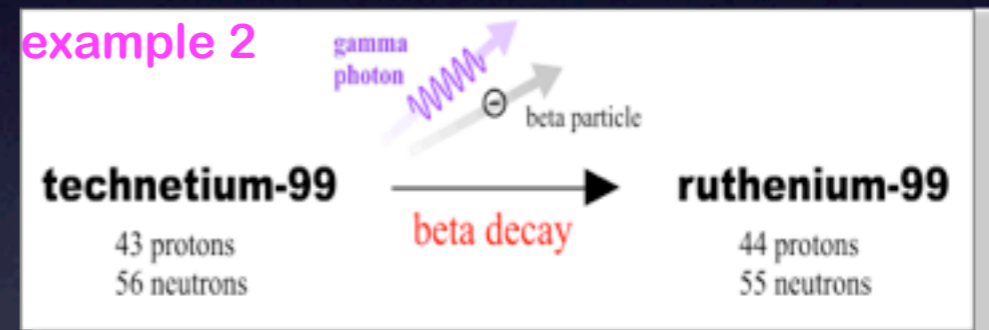
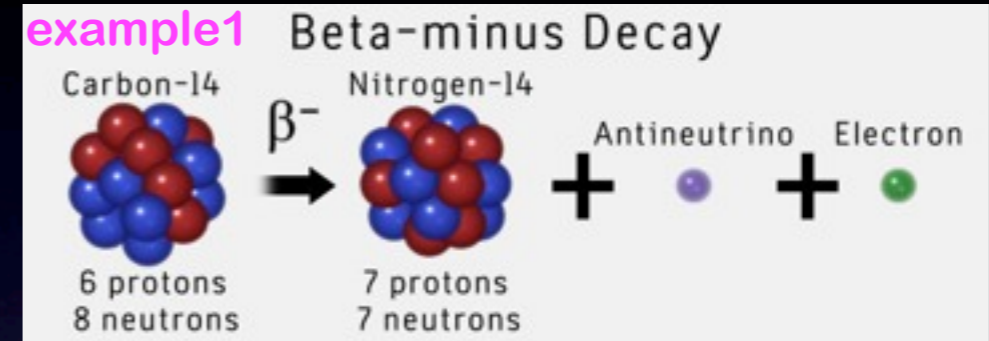
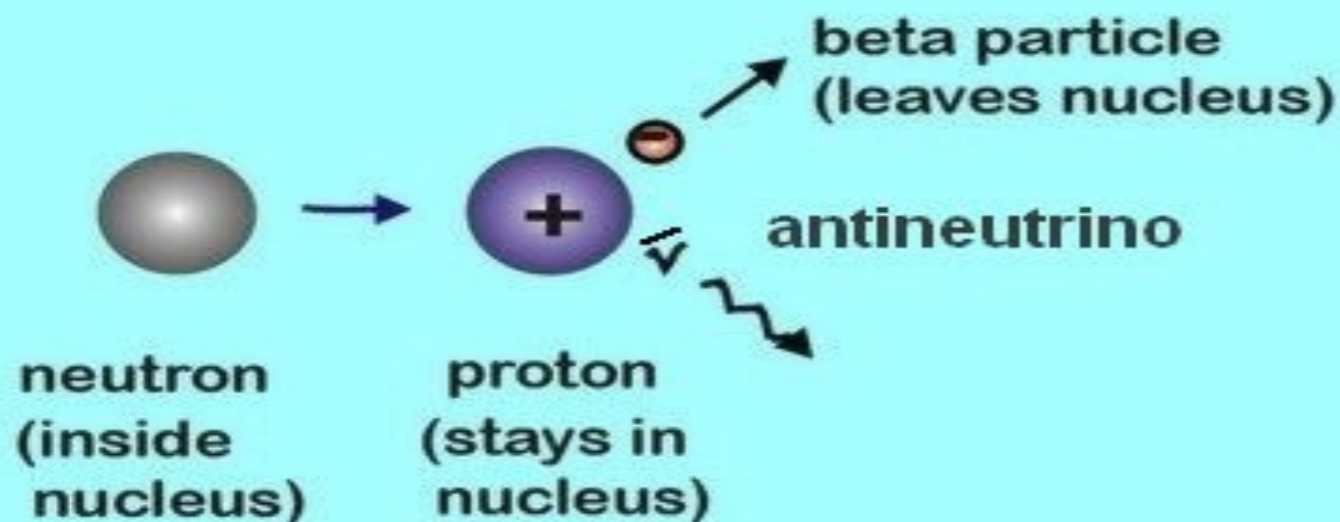
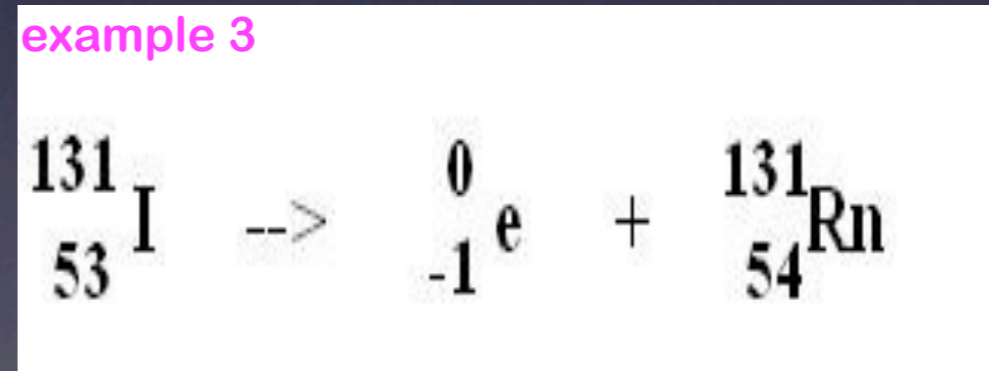


Figure 2



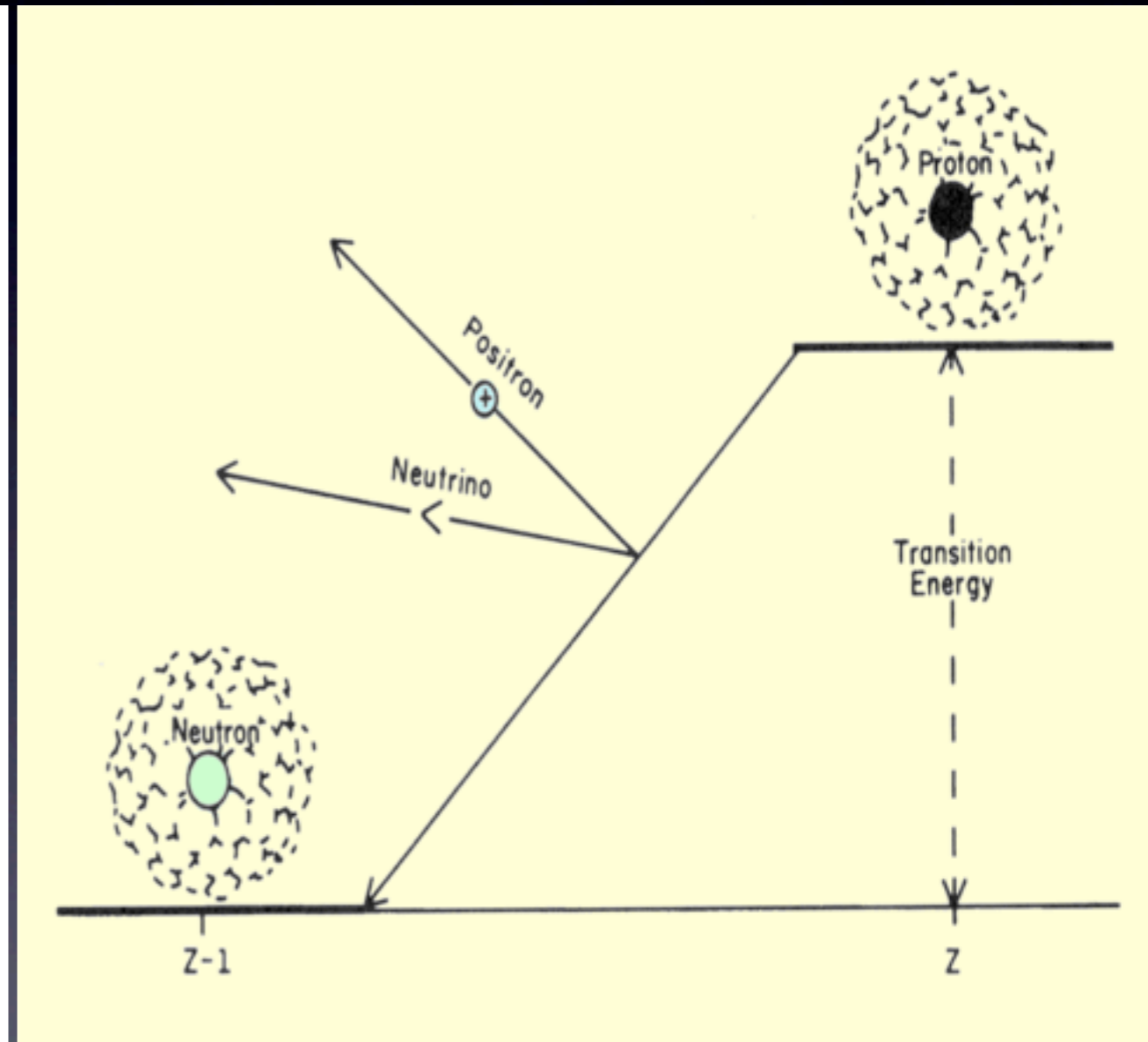
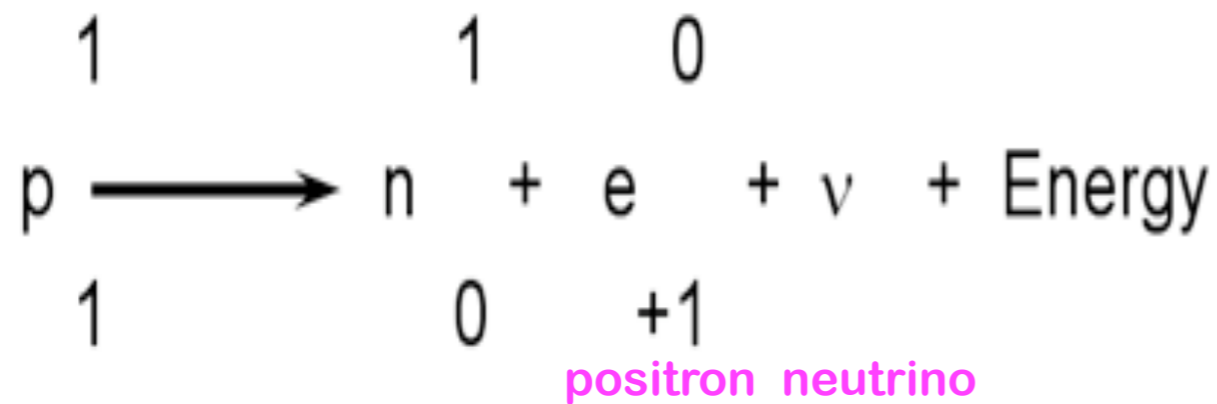
Radioactivity: beta positive particle or positron



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Positron Decay

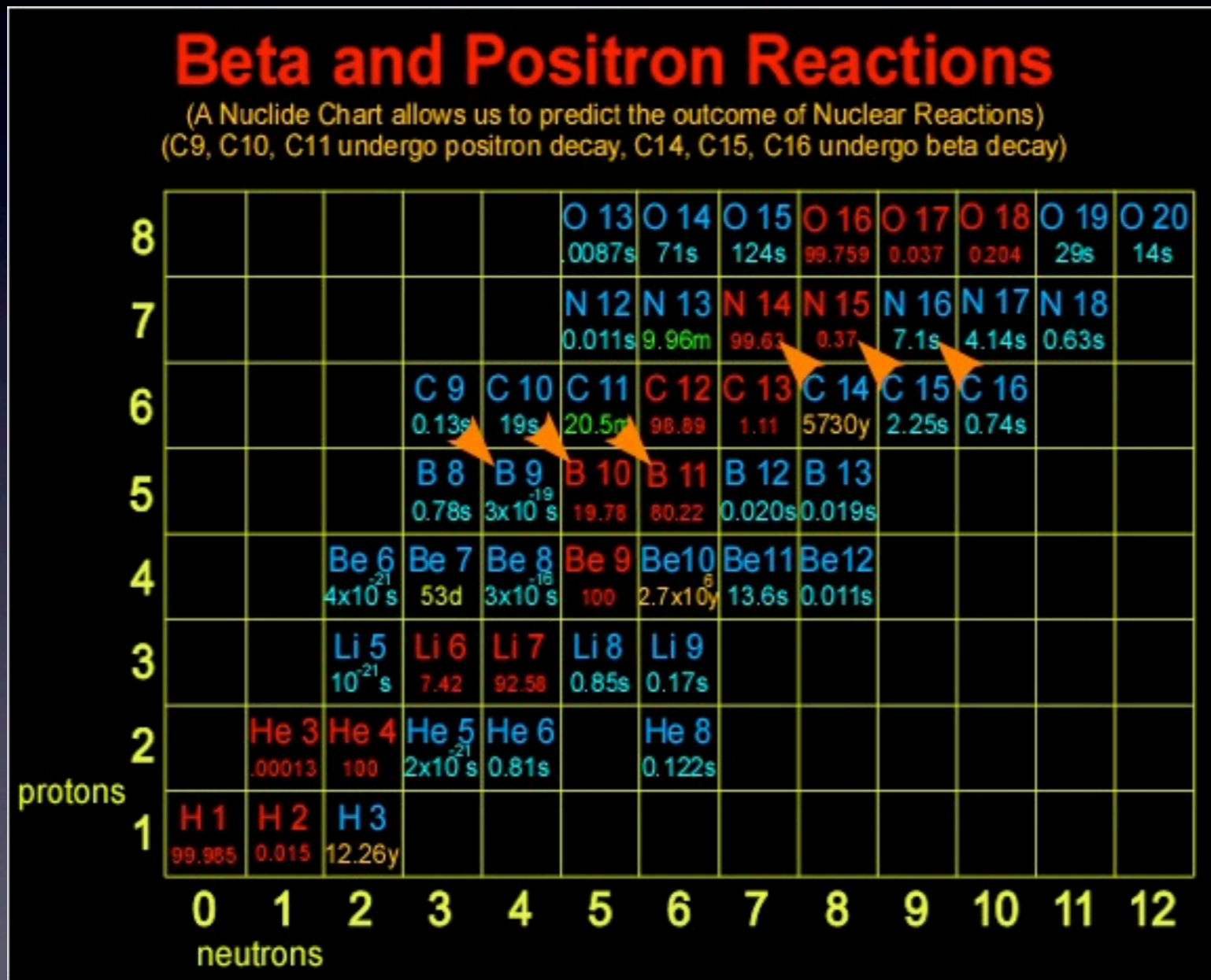
If the n/p ratio is too low, it may be increased by disintegration of a proton in the nucleus. This process is represented by the following equation:



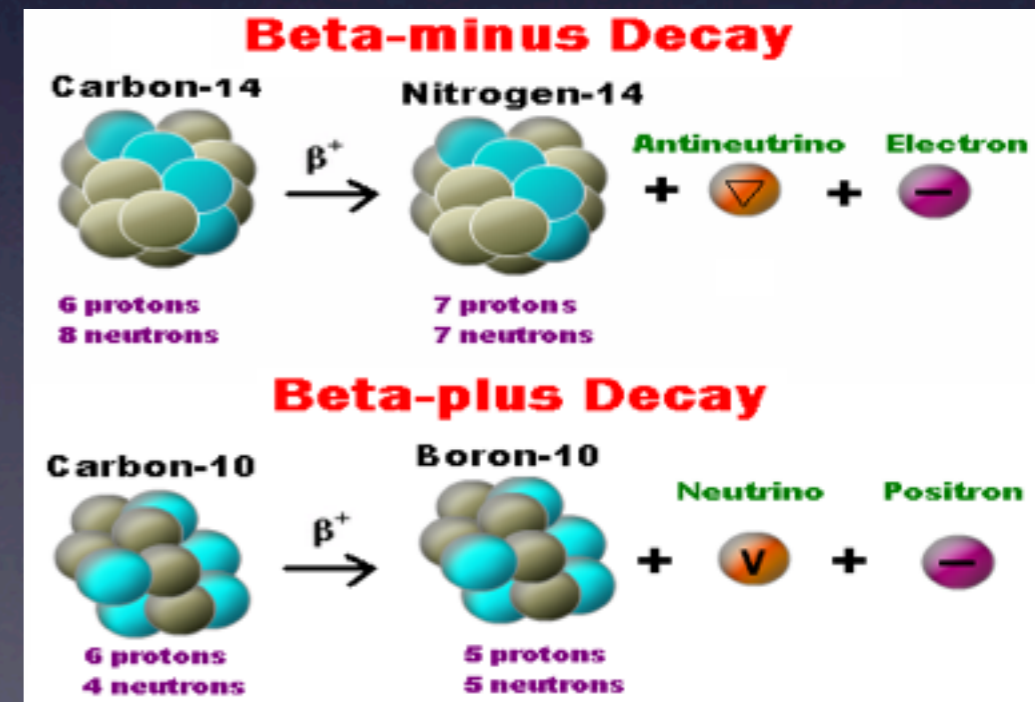
Beta and Positron Reactions: how to predict the outcome



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- Ratio Protons/Neutrons is determinant. Nucleus tend to lose excess particle.
- if N>P then N-> P (beta - decay) see C14, C15 and C16
- if P>N then P -> N aka beta+(positron) decay see C9,C10,C11



BETA RADIOACTIVITY: recap



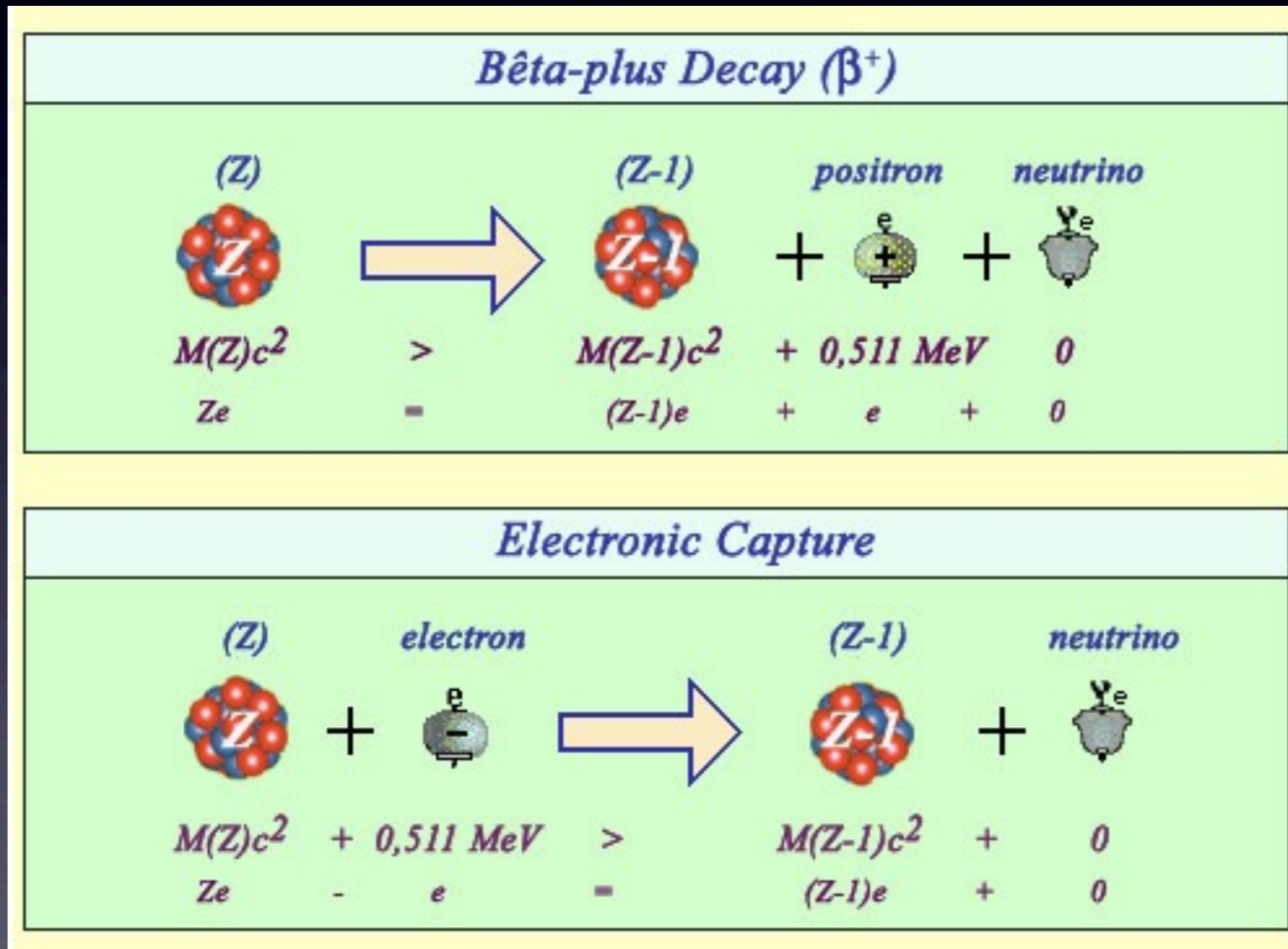
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BETA +	BETA -
if excess P then $P \rightarrow N$	if excess N then $N \rightarrow P$
W^+ boson carries weak force that produces this	W^- boson
positron and neutrino = final products	electron and antineutrino
O, F used in medicine in PET	many elements more common than beta+



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Weak Force - Beta Radioactivity particularity : electron capture



- Free energy is negative, $\Delta G < 0$, for spontaneous reactions to occur (thermodynamics); massive E is released when nuclear bonds are broken : $E=mc^2$ and here $\Delta G = \Delta E$, so for spontaneous decay : $\Delta E < 0$, $\Delta m < 0$.
- in heavier elements, Δm is small and do not generate enough energy for $P \rightarrow N$, so nucleus capture an $-e$ from inner (K) shell, process known as K capture/inverse beta decay, no positron just neutrino freed